

Synthesis and characterization of mRNA cap analogs containing phosphorothioate substitutions that bind tightly to eIF4E and are resistant to the decapping pyrophosphatase DcpS

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ABSTRACT

Analogues of the mRNA cap are widely employed to study processes involved in mRNA metabolism as well as being useful in biotechnology and medicinal applications. Here we describe synthesis of six dinucleotide cap analogs bearing a single phosphorothioate modification at either the α , β , or γ position of the 5',5'-triphosphate chain. Three of them were also modified with methyl groups at the 2'-O position of 7-methylguanosine to produce anti-reverse cap analogs (ARCAs). Due to the presence of stereogenic P centers in the phosphorothioate moieties, each analog was obtained as a mixture of two diastereomers, D1 and D2. The mixtures were resolved by RP HPLC, providing 12 different compounds. Fluorescence quenching experiments were employed to determine the association constant (K_{AS}) for complexes of the new analogs with eIF4E. We found that phosphorothioate modifications generally stabilized the complex between eIF4E and the cap analog. The most strongly bound phosphorothioate analog (the D1 isomer of the β -substituted analog m⁷GppspG) was characterized by a K_{AS} that was more than fourfold higher than that of its unmodified counterpart (m⁷GpppG). All analogs modified in the γ position were resistant to hydrolysis by the scavenger decapping pyrophosphatase DcpS from both human and *Caenorhabditis elegans* sources. The absolute configurations of the diastereomers D1 and D2 of analogs modified at the α position (i.e., m⁷GpppsG and m^{2,2',2''}-O-GpppsG) were established as S_P and R_P , respectively, using enzymatic digestion and correlation with the S_P and R_P diastereomers of guanosine 5'-O-(1-thiodiphosphate) (GDP α S). The analogs resistant to DcpS act as potent inhibitors of in vitro protein synthesis in rabbit reticulocyte lysates.

Keywords: phosphorothioate cap analogs; ARCA; decapping enzymes; DcpS; binding affinity for eIF4E; resistance to enzymatic hydrolysis

INTRODUCTION

A distinctive feature of all eukaryotic cellular mRNAs is the presence of a "cap" structure at their 5' ends. The typical cap consists of 7-methylguanosine connected via a 5',5'-triphosphate bridge to the first nucleotide of the transcript. The cap plays a key role in a variety of cellular processes

related to mRNA metabolism including splicing, intracellular transport, translation, translational repression, and turnover (Furuichi and Shatkin 2000; Richter and Sonenberg 2005). Synthetic cap analogs have served as useful tools for elucidating these physiological processes. Most progress has occurred in studies on the initiation of protein synthesis, but, more recently, modified cap analogs have yielded new insights into the degradation of mRNA (Grudzien et al. 2006; Grudzien-Nogalska et al. 2007).

During the initiation of translation, the cap is specifically recognized by eukaryotic initiation factor 4E (eIF4E), which participates in the recruitment of ribosomes to

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mRNA by virtue of its binding to eIF4G. This process is the rate-limiting step for initiation of translation under normal circumstances (absence of virus infection, cellular stress, etc.) and plays a central role in the regulation of translation (Gingras et al. 1999). Structural requirements for the cap-eIF4E interaction were elucidated, in a large measure, by determining the binding affinities of variously modified mono- and dinucleotide cap analogs for eIF4E (McCubbin et al. 1988; Carberry et al. 1989; Niedzwiecka et al. 2002), and comparing their activity as inhibitors of cap-dependent translation in cell-free systems (Cai et al. 1999 and citations therein). The inhibitory potency of individual analogs generally correlates well with their binding affinities for eIF4E. One motivation for designing cap analog inhibitors of eIF4E with higher affinity is for potential applications in anti-cancer therapy, since eIF4E is overexpressed in many types of tumor cells and since elevated eIF4E levels selectively increase translation of mRNAs important in malignant transformation and metastasis (De Benedetti and Graff 2004).

Capped mRNAs have found broad application for in vitro protein synthesis as well as for obtaining expression of exogenous mRNAs in living cells. Synthetic dinucleotide cap analogs serve as useful tools to obtain 5'-capped mRNAs via transcription in vitro. This is achieved by transcribing DNA template with either a bacterial (Contreras et al. 1982) or bacteriophage (Konarska et al. 1984; Yisraeli and Melton 1989) RNA polymerase in the presence of all four NTPs and a cap dinucleotide such as m⁷GpppG. These polymerases initiate transcription with nucleophilic attack by the 3'-OH of the guanosine (Guo) moiety in m⁷GpppG on the α phosphate of the next nucleoside triphosphate specified by the DNA template. Unfortunately, attack can also occur by the 3'-OH of the m⁷Guo, producing one-third to one-half of the transcripts capped in a reverse orientation, i.e., Gpppm⁷GpNpN... instead of m⁷GpppGpNpN... (Pasquinelli et al. 1995). Such reverse-capped transcripts decrease the overall translational activity of the mRNA. This problem was overcome by the introduction of anti-reverse cap analogs (ARCAs) bearing either 3'-O-methyl, 3'-H, or 2'-O-methyl modification in the m⁷Guo moiety, ensuring 100% correct orientation (Stepinski et al. 2001; Peng et al. 2002; Jemielity et al. 2003). ARCA-capped mRNAs were shown to have higher translational efficiency in vitro (Stepinski et al. 2001; Jemielity et al. 2003) as well as in cultured cells (Grudzien et al. 2006). Moreover, it is possible to increase mRNA translational efficiency by introducing further modifications, e.g., extending the length of the 5',5'-bridge (Jemielity et al. 2003).

Although a great number of cap analogs with various biochemical properties have been synthesized and exhaustively characterized, there have been few reports about analogs that are resistant to enzymatic hydrolysis. In eukaryotic cells there are two types of decapping enzymes, differing in their biological function, substrate specificity,

and regioselectivity of hydrolysis (Coller and Parker 2004). One of them, DcpS, is involved in the 3'→5' mRNA degradation pathway. DcpS utilizes free m⁷GpppN dinucleotides as well as short capped mRNAs released after mRNA 3'-deadenylation and degradation by the exosome, but it is unable to cleave cap structures attached to long RNA chains (Liu et al. 2002). The enzyme cleaves the cap regioselectively between the β and γ phosphates of the 5',5'-triphosphate bridge to release m⁷GMP and a downstream (oligo)nucleotide. A second decapping enzyme, the Dcp1/Dcp2 complex, is involved in the 5'→3' mRNA degradation pathway, in which deadenylation is followed by decapping and subsequent exonucleolytic digestion in the 5'→3' direction. Dcp2, the catalytic subunit of Dcp1/Dcp2, cleaves the cap between the α and β phosphates to release m⁷GDP and a 5'-phosphorylated mRNA chain (van Dijk et al. 2002; Wang et al. 2002), which is exposed to digestion by the exonuclease Xrn1.

In previous work, we designed and synthesized a series of cap analogs selectively modified with a methylenebisphosphonate moiety at either the α/β or β/γ position of the 5',5'-triphosphate bridge, i.e., m⁷GppCH₂pG, m₂^{7,3'-O}GppCH₂pG, m⁷GpCH₂ppG, m₂^{7,3'-O}GpCH₂ppG, that were predicted to be resistant to either Dcp1/Dcp2 or DcpS (Kalek et al. 2006). Several of these analogs were resistant toward DcpS, which allowed us to determine their equilibrium association with hDcpS (Kalek et al. 2006). When analogs containing the ARCA modification were incorporated into mRNA, it was found that mRNA capped with m₂^{7,3'-O}GppCH₂pG, but not m₂^{7,3'-O}GpCH₂ppG, was resistant to Dcp2 in vitro and stabilized mRNA in cultured cells (Grudzien et al. 2006). However, the translational efficiency of m₂^{7,3'-O}GppCH₂pG-capped mRNA was diminished compared with the parent m₂^{7,3'-O}GpppG-capped mRNA. A possible explanation for the observed reduction in translation may be related to the lower binding affinity of eIF4E for m₂^{7,3'-O}GppCH₂pG compared with m₂^{7,3'-O}GpppG (Kalek et al. 2005).

In the present study, we sought to develop cap analogs that retained high affinity for eIF4E, yet were resistant to cleavage by pyrophosphatases. In particular, we investigated the phosphorothioate modifications in the cap triphosphate bridge (i.e., one of nonbridging phosphate oxygens replaced by sulfur). This is considered to be an almost perfect mimic of the phosphate moiety since it is isosteric, pseudoisoelectronic, and has a similar charge distribution and similar net charge at physiological pH (Eckstein 1985). Furthermore, phosphorothioate nucleotide analogs are often resistant to enzymatic cleavage and have been used as nonhydrolyzable substrate analogs.

RESULTS

We prepared a series of six novel phosphorothioate mRNA cap analogs ("S analogs") bearing a single phosphorothioate moiety at either the α , β , or γ positions of the triphosphate

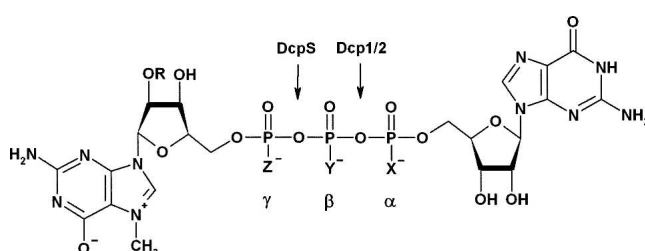
chain (Fig. 1). Due to the presence of a stereogenic P center, each S analog was obtained as a mixture of two diastereomers, designated D1 and D2 according to their elution order during reverse phase (RP) HPLC. Each S analog was successfully resolved by RP HPLC, providing 12 compounds that subsequently were characterized biochemically and biochemically. Six of the S analogs contained an ARCA modification, a 2'-O-methyl group in the m⁷Guo moiety, and hence, are termed S-ARCA. mRNAs capped with S-ARCA have been characterized with respect to enzymatic susceptibility to hDcp2 in vitro and both translational efficiency and stability in cultured cells (Grudzien-Nogalska et al. 2007). Notably, introduction of a phosphorothioate group at the β position produced resistance to Dcp2, increased half-life, and improved translational efficiency.

Synthesis of S analogs

The approach most commonly reported for the formation of a mixed phosphate-phosphorothioate anhydride bond has been coupling by the Michelson procedure employing diphenylphosphoryl chloride as an activating reagent (Eckstein and Goody 1976; Richard and Frey 1982). Although

this reaction is applicable for the synthesis of dinucleoside 5',5'-triphosphates, it has several features disadvantageous for phosphorothioate cap analog syntheses. The main difficulties would be poor solubility and stability of m⁷Guo-containing nucleotides in dimethyl formamide (DMF)/pyridine reaction mixtures. Also, low yields have been reported for analogs modified at the β position of the triphosphate chain due to instability of nucleoside 5'-O-(2-thiodiphosphates) in pyridine (Eckstein and Goody 1976; Ho and Frey 1984). Therefore, we attempted chemical synthesis of S analogs by a strategy similar to that originally developed for the analogs not modified in the 5',5'-polyphosphate bridge. In this strategy, two mononucleotide species, one of which has been previously converted to a reactive imidazolidine derivative, are coupled in DMF. The reaction is mediated by excess ZnCl₂, which significantly improves the solubility of reactants in organic media, prevents the hydrolysis of imidazolidine derivatives, and accelerates the reaction rate (Kadokura et al. 1997; Stepinski et al. 2001). In the final step of each synthesis leading to a particular S analog, a nucleotide imidazolidine derivative was coupled to either a nucleoside 5'-thiophosphate or 5'-(2-thiodiphosphate) in DMF in the presence of an excess of ZnCl₂.

The synthetic pathway leading to analogs 1 and 4 modified at the α position of the 5',5'-triphosphate bridge (i.e., m⁷Gppp_sG and m₂^{7,2'-O}Gppp_sG) is depicted in Figure 2. In both final coupling reactions, a 1.5- to twofold excess of phosphorimidazolidine was used to ensure complete consumption of the nucleoside 5'-thiophosphate. The coupling proceeded steadily, leading to almost complete consumption of the substrate within 1–2 d. The synthesis of analogs 3 and 6 modified at the γ position (i.e., m⁷Gp_sppG and m₂^{7,2'-O}Gp_sppG), which is depicted in Figure 3, was similar to the one described above. In each case, formation of two diastereomers was indicated by RP HPLC, as shown in Figure 4. The intermediate nucleoside 5'-thiophosphates 9–11 were synthesized via thiophosphorylation of appropriate nucleosides by PSCl₃ in trimethyl phosphate in the presence of 2,6-dimethylpyridine at 0°C, similar to the previously reported procedures (Moran and Whitesides 1984). In the case of compounds 10 and 11, the methylation at the N7 position had to be performed at the nucleoside stage, before the thiophosphorylation step; otherwise, methyl iodide preferably alkylates the sulfur atom (data not shown). Conversion of nucleoside 5'-diphosphates into their imidazolidine derivatives (7, 8, and 12–15) was easily achieved via reaction with imidazole employing the 2,2'-dithiodipyridine/triphenylphosphine activation system (Mukaiyama and Hashimoto 1971). The analogs modified at the β position, i.e., m⁷Gpp_spG (2) and m₂^{7,2'-O}Gpp_spG (5), were synthesized as depicted in Figure 5. HPLC analysis of the final coupling revealed formation of two P diastereoisomers. However, their retention times were very similar. To obtain the intermediate nucleoside



Compound	Abbreviation	X	Y	Z	R	Configuration
1a	m ⁷ Gppp _s G (D1)	S	O	O	H	S _p
1b	m ⁷ Gppp _s G (D2)	S	O	O	H	R _p
2a	m ⁷ Gpp _s pG (D1)	O	S	O	H	n.a.
2b	m ⁷ Gpp _s pG (D2)	O	S	O	H	n.a.
3a	m ⁷ Gp _s ppG (D1)	O	O	S	H	n.a.
3b	m ⁷ Gp _s ppG (D2)	O	O	S	H	n.a.
4a	m ₂ ^{7,2'-O} Gppp _s G (D1)	S	O	O	CH ₃	S _p
4b	m ₂ ^{7,2'-O} Gppp _s G (D2)	S	O	O	CH ₃	R _p
5a	m ₂ ^{7,2'-O} Gpp _s pG (D1)	O	S	O	CH ₃	n.a.
5b	m ₂ ^{7,2'-O} Gpp _s pG (D2)	O	S	O	CH ₃	n.a.
6a	m ₂ ^{7,2'-O} Gp _s ppG (D1)	O	O	S	CH ₃	n.a.
6b	m ₂ ^{7,2'-O} Gp _s ppG (D2)	O	O	S	CH ₃	n.a.

n.a. - not assigned

FIGURE 1. Structures of the phosphorothioate cap analogs synthesized in this study. The respective DcpS and Dcp1/Dcp2 cleavage sites in the cap structure are marked together with α/β/γ phosphate designations.

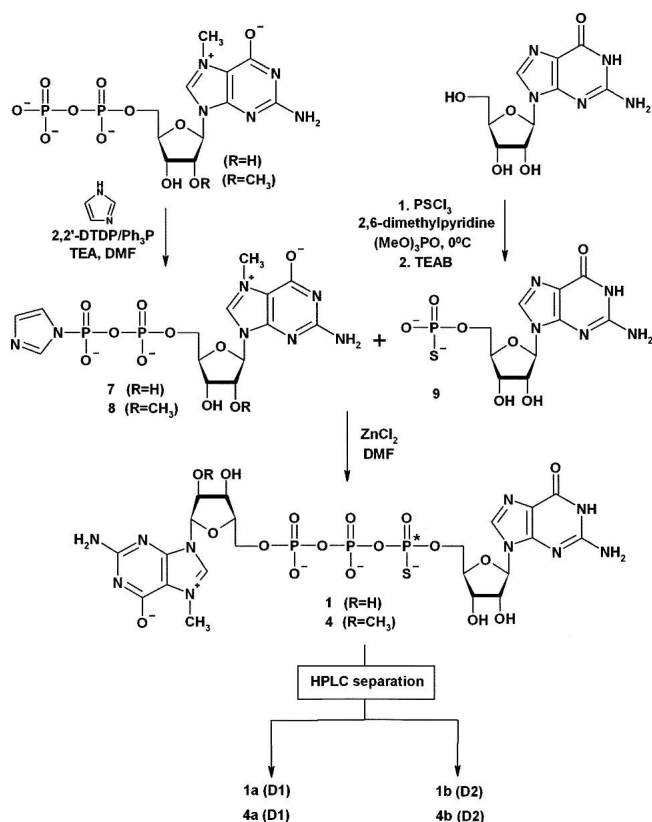


FIGURE 2. Synthesis of cap analogs modified at the α position of the 5',5'-triphosphate bridge (compounds 1 and 4).

5'-O-(2-thiodiphosphates) 16 and 17, we employed a recently developed coupling reaction between a nucleoside 5'-monophosphate imidazolide and thiophosphate (PSO_3^{3-}) triethylammonium salt as a nucleophile (Kowalska et al. 2007).

In all reactions leading to cap analogs 1–6, HPLC analysis revealed that the desired compounds were formed as the major products, with only moderate amounts of by-products. Nonetheless, preparative yields were surprisingly lower than those indicated by HPLC, being in the range of 10%–20% overall. This is probably due to large losses of material during lengthy separation of diastereoisomers by RP HPLC, which in many cases was performed repeatedly in order to obtain diastereomerically pure samples.

Binding affinities for eIF4E

We determined the binding affinities of S analogs for murine eIF4E by fluorescence quenching (Fig. 6). The descending titration curves arise from the quenching of intrinsic Trp fluorescence in eIF4E upon cap binding, whereas the increasing fluorescence signals after reaching the plateau originate from fluorescence emission of unbound cap analog. The K_{AS} values and free energies of binding (ΔG°) of the S analogs are presented in Table 1, together with the same data for their unmodified parent

compounds. Surprisingly, the presence of the phosphorothioate moiety does not reduce binding affinity for eIF4E, instead in most cases it significantly increases it. The K_{AS} values are strongly dependent both on the position of the phosphorothioate modification and on the absolute configuration of the asymmetric P center. Interestingly, in each pair of diastereomers, the D1 member binds to eIF4E with an affinity that is 2.3- to 4.5-fold higher than the D2 member or the parent analog (Fig. 7). For instance, K_{AS} for the D1 isomer of $m^7\text{Gp}_{\text{Spp}}\text{G}$ is threefold higher than for D2 or $m^7\text{GpppG}$. Similarly, K_{AS} for the D1 isomer of $m^7\text{Gpp}_{\text{S}}\text{pG}$, is twofold higher than for D2 and 4.5-fold higher than for $m^7\text{GpppG}$. The S-ARCA series shows similar corresponding differences. The greatest differences in binding affinities between the D1/D2 diastereomers were observed for the γ -modified analogs, whereas the greatest differences between modified and unmodified pairs were observed for β -substituted analogs.

Susceptibility to enzymatic hydrolysis by DcpS

We subjected the new series of S analogs to in vitro enzymatic hydrolysis catalyzed by DcpS from both human and *Caenorhabditis elegans* sources. In all experiments, the

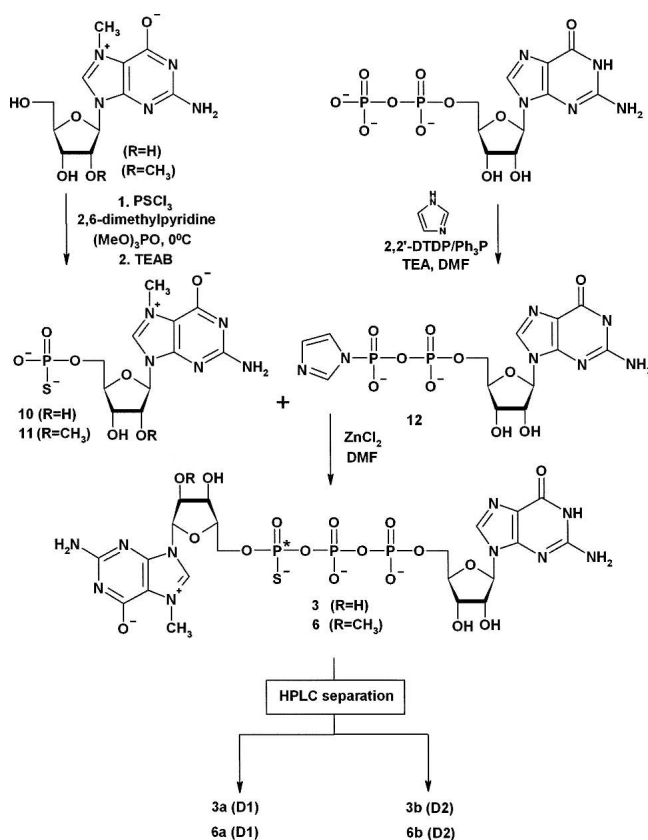


FIGURE 3. Synthesis of cap analogs modified at the γ position of the 5',5'-triphosphate bridge (compounds 3 and 6).

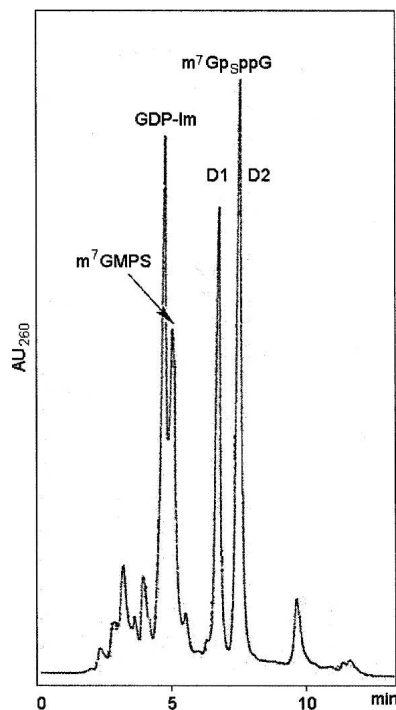


FIGURE 4. RP HPLC profile of coupling reaction between 7-methylguanosine 5'-O-thiophosphate (m^7 GMPs) and guanosine 5'-O-diphosphate imidazolide derivative (GDP-Im) leading to two diastereomers of m^7 Gp₅ppG (3a and 3b) marked D1 and D2 according to their elution order.

corresponding unmodified cap analog was used as a positive control, i.e., m^7 GpppG for non-ARCA S analogs and $m_2^{7,2'-O}$ GpppG for S-ARCA. The amount of DcpS enzyme was optimized to provide complete degradation of the control substrate within 40–90 min. The samples collected from reaction mixtures at various time intervals were analyzed by RP HPLC (as described in Materials and Methods). We found that S analogs modified at the γ position were resistant to hydrolysis, independent of the P-center absolute configuration (Table 2). Identical results were obtained with longer reaction times up to 24 h, different concentrations of enzyme, and different reaction buffers. All other S analogs were hydrolyzed by hDcpS with efficiencies comparable to the unmodified parent analog. No significant differences were observed for S-analog hydrolysis by DcpS from human and *C. elegans* sources.

Analysis of the DcpS degradation products of analogs modified at the α position allowed us to determine their absolute configuration of the asymmetric P centers. We found that hydrolysis of either m^7 Gppp₅G (D1) or m^7 Gppp₅G (D2) by DcpS leads to m^7 GMP and either the D1 or D2 isomer of guanosine 5'-O-(1-thiodiphosphate) (GDP α S), whereas hydrolysis of either $m_2^{7,2'-O}$ Gppp₅G (D1) or $m_2^{7,2'-O}$ Gppp₅G (D2) leads to $m_2^{7,2'-O}$ GMP and either the D1 or D2 isomer of GDP α S (Fig. 8). The absolute configuration of GDP α S diastereomers is known—the

isomer D1, which elutes earlier on RP HPLC, has been assigned the S_P configuration (Connolly et al. 1982). Since there is no possibility of configurational change at the α -phosphorus atom during the enzymatic hydrolysis, we thereby assigned the configuration of m^7 Gppp₅G (D1) and $m_2^{7,2'-O}$ Gppp₅G (D1) as S_P .

DcpS-resistant cap analogs as inhibitors of cap-dependent translation

The ability of the new S analogs to inhibit cap-dependent translation was assayed in a rabbit reticulocyte lysate system programmed with natural rabbit globin mRNA. Of the 12 S analogs, we selected two that were modified at the γ position, m^7 Gp₅ppG (D1) and m^7 Gp₅ppG (D2), since they were found to be resistant toward DcpS and thus are potentially more stable in vivo. Representative data for the two S analogs and the unmodified parent compound, m^7 GpppG, are shown in Figure 9. Data for inhibition of translation were fit to a theoretical curve that describes cap-dependent translation as a function of a competitive inhibitor of mRNA binding (Cai et al. 1999). This allowed us to determine K_I , the cap analog concentration at which cap-dependent translation is inhibited by 50% (Table 3). Both S analogs were found to be better inhibitors of

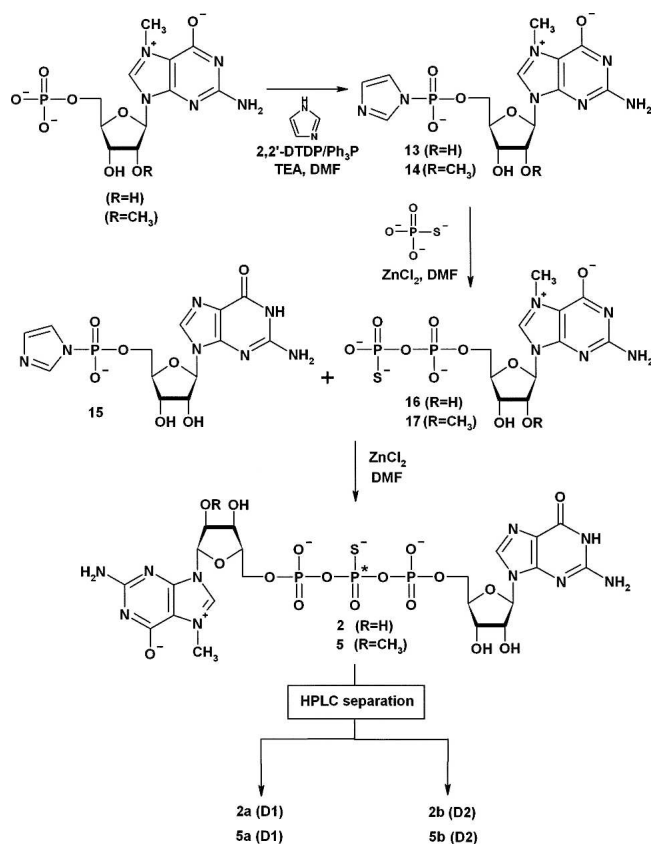


FIGURE 5. Synthesis of cap analogs modified at the β position of the 5',5'-triphosphate bridge (compounds 2 and 5).

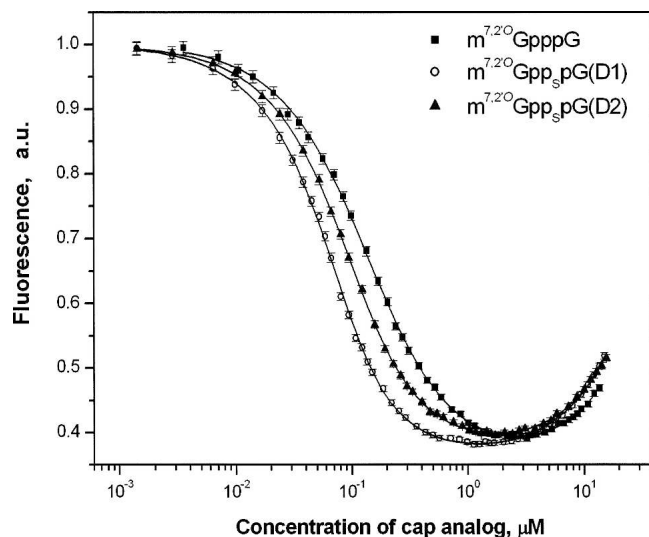


FIGURE 6. Fluorescence titration curves for binding of $m^{7,2'-O}$ GpppG, $m^{7,2'-O}$ GppspG (D1), and $m^{7,2'-O}$ GppspG (D2) to mouse eIF4E (28–217). The observed shift of curves toward the higher ligand concentration indicated a weaker binding between eIF4E and the corresponding cap analog. The observed increasing fluorescence signal at a higher concentration of ligands originates from the emission of free cap analog in solution. The intensity of fluorescence is represented as relative values. (a.u.) Arbitrary units.

cap-dependent translation than m^7 GpppG, which constitutes additional evidence that the phosphorothioate moiety generally strengthens the cap–eIF4E interaction. Moreover, m^7 GpsspG (D1) was significantly more inhibitory than its D2 counterpart ($K_i = 4.1 \pm 0.2 \mu\text{M}$ versus $K_i = 12.1 \pm 3.2 \mu\text{M}$), which is in agreement with its higher binding affinity for eIF4E based on fluorescent titration analyses ($K_{AS} = 30.8 \pm 0.5 \mu\text{M}^{-1}$ versus $K_{AS} = 10.0 \pm 0.2 \mu\text{M}^{-1}$).

DISCUSSION

The series of new cap analogs 1–6 with phosphorothioate modifications within the triphosphate bridge is the next step in our search for cap analogs with increased resistance to enzymatic cleavage. Although the overall strategy for the multistep synthesis leading to S analogs was similar to those reported previously (Stepinski et al. 2001; Jemielity et al. 2003), we demonstrated for the first time the use of nucleoside 5'-O-phosphorothioates and 5'-O-(2-thiodiphosphates) as nucleophiles in ZnCl_2 -promoted coupling reactions with nucleotide imidazolide derivatives. The S analogs are also the first examples of P-chiral cap analogs. For each S analog we found conditions to

resolve the P diastereomers by RP HPLC, which allowed us to use diastereomerically pure samples in the subsequent biophysical and biochemical studies.

The determination of the binding affinity for eIF4E (expressed as an association equilibrium constant, K_{AS}) is a reliable biophysical test enabling one to predict the biochemical properties of newly synthesized cap analogs. Generally, higher K_{AS} values correlate positively with stronger inhibitory potential of the cap analog and higher translational efficiency when incorporated into mRNA. As mentioned in the Introduction, the methylenebisphosphonate modification of the triphosphate bridge reduced the affinity of cap analogs for eIF4E, which reduced the translational efficiency of mRNAs capped with these analogs (Grudzien et al. 2006). By contrast, the phosphorothioate modification present at any position of the triphosphate bridge does not diminish the cap–eIF4E interaction. The differences in binding affinity of different S analogs for eIF4E can be explained by the geometry of interactions within the cap-binding pocket (Fig. 10). The cap–eIF4E complex, besides the stacking of 7-methylguanine between Trp-56 and Trp-102, is stabilized by hydrogen bonds and salt bridges between the negatively charged triphosphate bridge and positively charged amino acid residues in the cap binding pocket. Biophysical studies of eIF4E binding with cap analogs have shown that the addition to m^7 Guo of γ , β , and α phosphates results in a progressive increase in binding affinity for eIF4E, with corresponding energetic contributions ($\Delta\Delta G^\circ$) of ~ 3.0 , ~ 1.9 , and ~ 0.9 kcal/mol, respectively (Niedzwiecka et al. 2002). This observation is in agreement with data from X-ray crystal structures that reveal two salt bridges crucial for cap–eIF4E interaction, one between the nonbridging pro- R_P oxygen at the γ position and the guanidinium group of Arg-157, and another between the nonbridging

TABLE 1. Equilibrium association constants (K_{AS}) and binding free energies (ΔG°) for the binding of murine eIF4E (28–217) to phosphorothioate cap analogs, as determined by fluorescence quenching

Compound	Cap analog	$K_{AS} (\mu\text{M}^{-1})$	ΔG° (kcal/mol)
1a	m^7 GpppG	9.4 ± 0.4	-9.35 ± 0.02
	m^7 GpppsG (D1)	23.6 ± 0.8	-9.88 ± 0.02
	m^7 GpppsG (D2)	13.1 ± 0.8	-9.54 ± 0.03
2a	m^7 GppspG (D1)	45.0 ± 1.1	-10.26 ± 0.01
2b	m^7 GppspG (D2)	23.0 ± 0.4	-9.87 ± 0.01
3a	m^7 GpsspG (D1)	30.8 ± 0.5	-10.04 ± 0.01
3b	m^7 GpsspG (D2)	10.0 ± 0.2	-9.39 ± 0.01
4a	$m_2^{7,2'-O}$ GpppG	10.8 ± 0.3	-9.43 ± 0.02
	$m_2^{7,2'-O}$ GpppsG (D1)	19.2 ± 0.8	-9.76 ± 0.03
	$m_2^{7,2'-O}$ GpppsG (D2)	15.0 ± 0.6	-9.62 ± 0.02
5a	$m_2^{7,2'-O}$ GppspG (D1)	43.1 ± 1.4	-10.23 ± 0.02
5b	$m_2^{7,2'-O}$ GppspG (D2)	19.3 ± 2.2	-9.77 ± 0.07
6a	$m_2^{7,2'-O}$ GpsspG (D1)	35.2 ± 1.1	-10.12 ± 0.02
6b	$m_2^{7,2'-O}$ GpsspG (D2)	12.9 ± 0.4	-9.53 ± 0.02

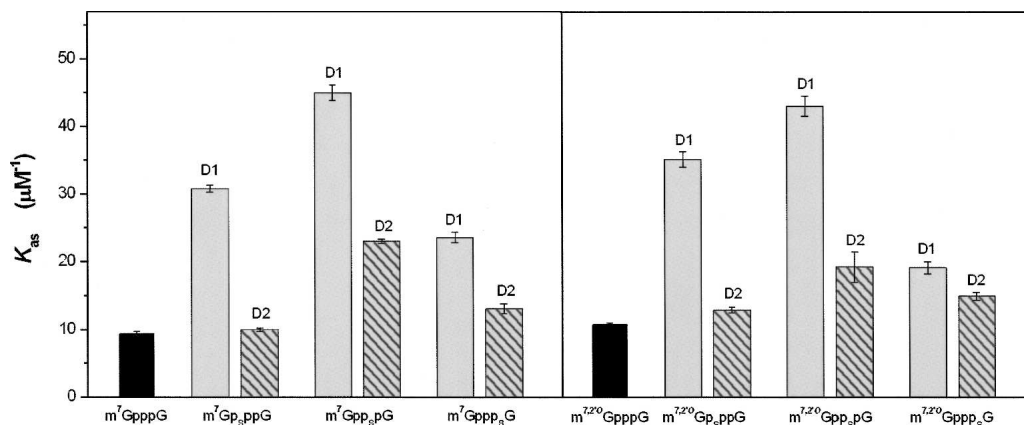


FIGURE 7. Comparison of association constants for complexes between eIF4E and the standard cap analog (m⁷GpppG) or cap S analogs. The influence of sulfur presence in the phosphate chain of the cap analog on its binding by eIF4E depends strongly on its position and the absolute configuration of asymmetric P centers.

pro-*R_P* oxygen at the β position and the ammonium group of Lys-162. Four other oxygens are involved in H-bond interactions, either directly or mediated by water molecules. These data are convergent with our finding that the largest increases in the binding affinity were observed for one of the diastereomers substituted at the γ position and one substituted at the β-position [m⁷Gp_sppG (D1), **3a**, and m⁷Gpp_spG (D1), **2a**; with corresponding observations in the S-ARCA series]. Based on known physicochemical properties of phosphorothioate anions, we propose that the effect of the oxygen-to-sulfur substitution results from increased polarizability of phosphorothioate in comparison to the phosphate group. It was previously suggested that binding of phosphorothioates at protein active sites containing ammonium and guanidinium moieties is more similar to crystalline phosphorothioate salts than to water-solvated species (Baraniak and Frey 1988). In phosphorothioate crystalline salts, positively charged counterions coordinate closely to oxygen, drawing the electron density from sulfur. This results in shortening of the P–S bond and decreasing of the P–O bond order (Saenger and Eckstein 1970), unlike water-solvated species where the P–S bonds orders are close to one (Frey and Sammons 1985). Therefore, most likely the greatest increase in the binding affinity of eIF4E for phosphorothioate cap analogs is observed if the sulfur atom occupies either γ-pro-*S_P* or β-pro-*S_P* position in the triphosphate bridge, allowing a shift of the negative charge toward the respective pro-*R_P* oxygen involved in the salt-bridge interaction. In this case, m⁷Gp_sppG (D1) and m⁷Gpp_spG (D1), as well as their ARCA counterparts, would have the *S_P* configuration. However, our hypothesis cannot be properly verified without knowledge of the absolute configuration at the asymmetric P centers of the β- and γ-substituted S analogs. Unfortunately, we have been able to assign the absolute configuration only for the analogs substituted at the α position. However, our observation of the upfield shift of the m⁷Guo

H8 proton in the ¹H NMR spectrum of m⁷Gp_sppG (D1) and m^{2,2',2''}-OGp_sppG (D1), together with the previously reported correlation between the retention time on the RP HPLC column and the absolute configurations of nucleoside 5'-O-(1-thiotriphosphates) and 5'-O-(1-thiodiphosphates) (i.e., *S_P* is the faster) (Ludwig and Eckstein 1989; Li et al. 2005), suggest that the D1 isomers are also of the *S_P* configuration in the case of γ-substituted S analogs.

TABLE 2. Susceptibility of S analogs to enzymatic hydrolysis by DcpS (from human and *C. elegans*) in vitro

Compound	Cap analog	Reactivity
1a	m ⁷ GpppG	Hydrolyzed
	m ⁷ GpppsG (D1)	Hydrolyzed
1b	m ⁷ GpppsG (D2)	Hydrolyzed
2a	m ⁷ GppspG (D1)	Hydrolyzed
2b	m ⁷ GppspG (D2)	Hydrolyzed
3a	m ⁷ GpsppG (D1)	Resistant
3b	m ⁷ GpsppG (D2)	Resistant
4a	m ^{2,2',2''} -OGpppG	Hydrolyzed
	m ^{2,2',2''} -OGpppsG (D1)	Hydrolyzed
4b	m ^{2,2',2''} -OGpppsG (D2)	Hydrolyzed
5a	m ^{2,2',2''} -OGppspG (D1)	Hydrolyzed
5b	m ^{2,2',2''} -OGppspG (D2)	Hydrolyzed
6a	m ^{2,2',2''} -OGpsppG (D1)	Resistant
6b	m ^{2,2',2''} -OGpsppG (D2)	Resistant

Cap analogs at 4 μM concentration were subjected to enzymatic digestion by DcpS in conditions leading to complete degradation of the unmodified parent compound (i.e., m⁷GpppG for non-ARCA S analogs and m^{2,2',2''}-OGpppG for ARCA). Samples collected from reaction mixtures at various time intervals were analyzed by RP HPLC with UV detection at 260 nm as described in the Materials and Methods. The analogs assigned as *resistant* remained completely undigested under the applied conditions, whereas the analogs assigned as *hydrolyzed* were hydrolyzed by DcpS with efficiency comparable to the respective unmodified parent compound.

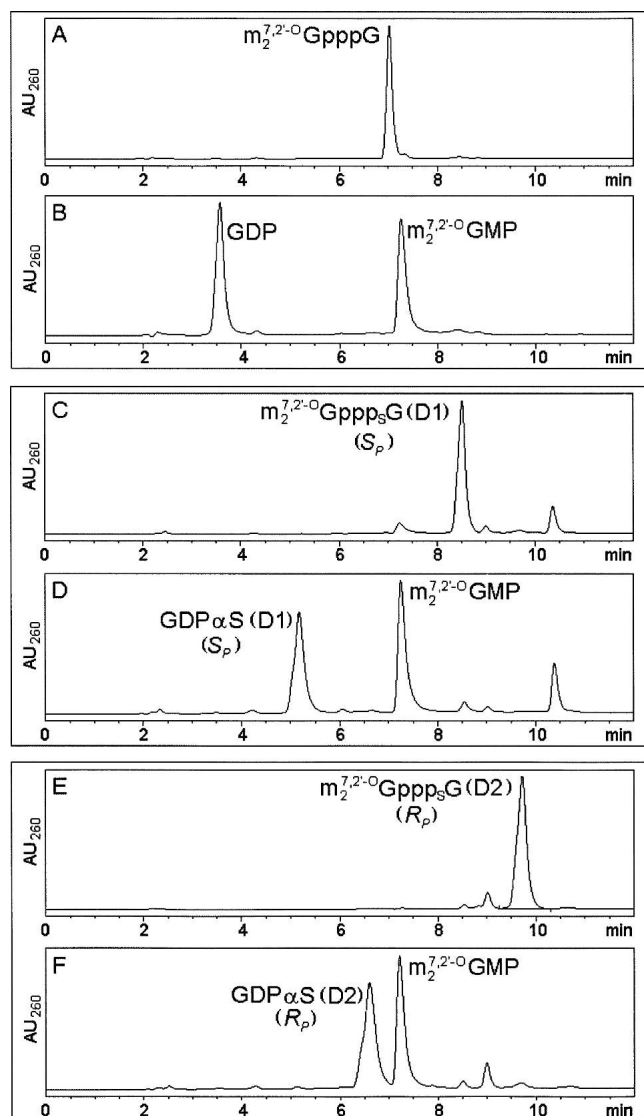


FIGURE 8. Correlation of the absolute configuration of D1 and D2 isomers of $m_2^{7,2'-O}$ GpppS (compounds **4a** and **4b**) with S_P and R_P diastereomers of guanosine 5'-O-(2-thiodiphosphate) (GDP α S). The RP HPLC analysis shows that unmodified ARCA analog $m_2^{7,2'-O}$ GpppG (A) is hydrolyzed by DcpS to $m_2^{7,2'-O}$ GMP and GDP (B). The D1 isomer of $m_2^{7,2'-O}$ GpppS (C) is hydrolyzed by DcpS to $m_2^{7,2'-O}$ GMP and the D1 isomer of GDP α S (D), whereas the D2 isomer of $m_2^{7,2'-O}$ GpppS (E) is hydrolyzed by DcpS to $m_2^{7,2'-O}$ GMP and the D2 isomer of GDP α S (F). The absolute configuration of the D1 and D2 isomers of GDP α S was assigned previously as S_P and R_P , respectively.

As determined by in vitro enzymatic hydrolysis experiments, the resistance to DcpS was observed only for the S analogs modified at the γ position. This is the position that is directly involved in the catalytic mechanism. DcpS, as a member of the HIT family of pyrophosphatases, employs a histidine triad motif to carry out the hydrolysis (Liu et al. 2002). The hDcpS catalysis proceeds by the nucleophilic attack of the His 277 imidazole ring on the cap γ phosphate

(Gu et al. 2004), and the mechanism presumably involves a trigonal-bipyramidal pentacoordinated transition state (Lima et al. 1997). Since the DcpS cap-binding pocket is relatively tight, the resistance of the γ -substituted S analogs to DcpS is likely due to steric bulk of the sulfur atom in comparison to oxygen, hindering the course of nucleophilic substitution. Although it was previously found that DcpS orthologs from human and *C. elegans* differ in their substrate specificity (*C. elegans*, but not human, DcpS can hydrolyze $m_3^{2,2,7}$ GpppG cap and requires shorter capped oligonucleotides as a substrate) (Cohen et al. 2004), we have not found any significant differences in their ability to hydrolyze S analogs, indicating that the general catalytic mechanism for these two enzymes is similar.

The finding that both DcpS-resistant S analogs, m^7 GpsppG (D1) and (D2), inhibit cap-dependent translation in vitro more effectively than m^7 GpppG contributes additional evidence that the phosphorothioate moiety does not disturb, but rather increases, the strength of the cap-eIF4E interaction. The differences in K_I between the diastereomers were in qualitative correlation with their binding affinities for eIF4E (Table 3). Notably, the K_I value for m^7 GpsppG (D1) is comparable to that of m^7 GTP ($K_I = 4.4 \mu\text{M}$) (Cai et al. 1999), which makes it by far the most inhibitory dinucleoside triphosphate analog developed to date.

The substrate specificity of DcpS suggests that the major role of this enzyme is removal of mRNA degradation products, which could interact with cap-binding proteins, interfering with crucial cellular processes and thus becoming toxic to the cell. In this respect, DcpS activity poses a putative threat to synthetic cap analogs destined for studies

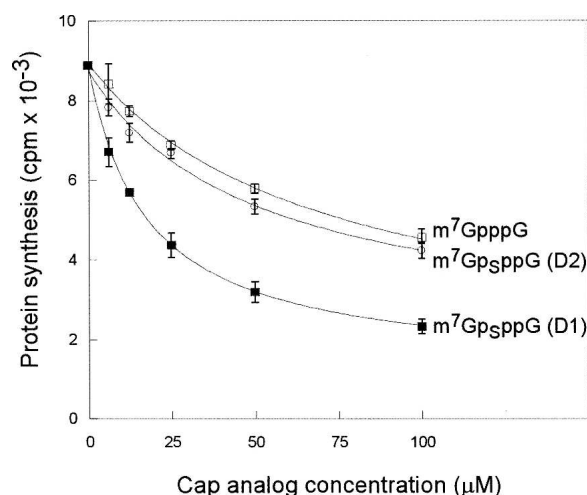


FIGURE 9. Inhibition of globin mRNA translation in rabbit reticulocyte lysate by m^7 GpppG ($\square$), m^7 GpsppG (D1) ($\blacksquare$), and m^7 GpsppG (D2) ($\circ$). Natural rabbit globin mRNA was translated at $5 \mu\text{g/mL}$ in a rabbit reticulocyte lysate system, and globin synthesis was detected by incorporation of [^3H] Leu into protein as described in the Materials and Methods.

TABLE 3. Inhibitory constants (K_i) for inhibition of cap-dependent translation by γ -modified S- analogs in a rabbit reticulocyte lysate translation system

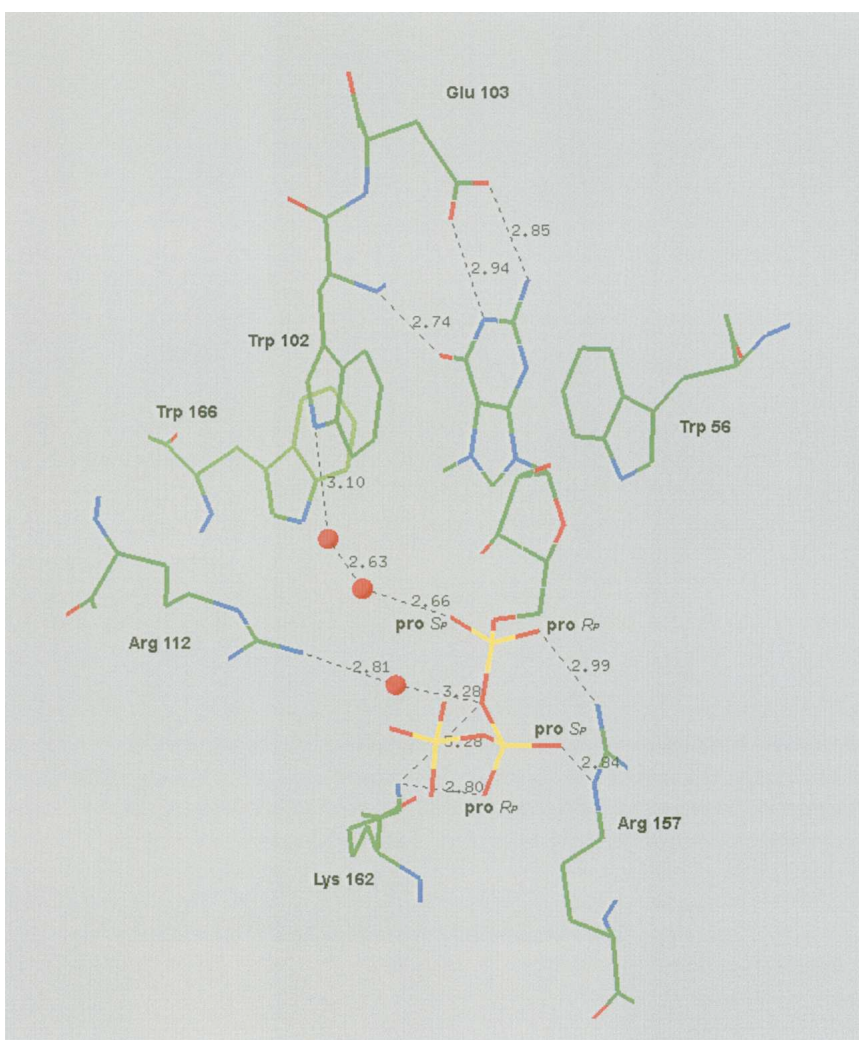
Compound	Cap analog	K_i [μ M]
	m^7GpppG	17.1 ± 2.5
3a	$m^7GpsppG$ (D1)	4.1 ± 0.2
3b	$m^7GpsppG$ (D2)	12.1 ± 3.2

in vivo, e.g., those with potential therapeutic value. Many studies have shown that eIF4E is an attractive target for anti-cancer directed therapy. The ability of cap analogs to inhibit translation in vitro suggests that highly specific cap analog inhibitors of eIF4E might counteract elevated eIF4E levels in tumor cells. However, the inhibition of cap-dependent translation by cap analogs in vivo has not yet been demonstrated, mainly because such an application requires that cap analogs not only have strong inhibitory potency but also retain sufficiently high stability. One of our motivations for modifying the 5',5'-phosphate bridge was to produce a cap analog resistant to DcpS hydrolysis that would inhibit translation through its interaction with eIF4E. The two DcpS-resistant S analogs, $m^7GpsppG$ (D1) and (D2), offer a promising new solution to this problem.

Whereas DcpS is responsible for degradation of free cap structures and capped mRNA degradation products, the decapping and thus degradation of intact mRNAs is dependent on Dcp1/Dcp2 activity. Protecting mRNA from Dcp2 cleavage by introducing an appropriately chemically modified cap analog at its 5'-end should produce transcripts with increased stability. In a related study (Grudzien-Nogalska et al. 2007), we demonstrated that phosphorothioate modification at the β position protects mRNAs against Dcp2 in vitro. This results in prolonged half-lives and higher translational efficiencies of such mRNAs in cultured mammalian cells. The ability to produce more stable mRNAs with higher translational efficiency can be advantageous for applications exploiting the concept of RNA-based gene delivery. For instance, transfection with

mRNAs encoding tumor-associated antigens is a promising approach to cancer therapy by autoimmunization with a patient's own dendritic cells (Mitchell and Nair 2000). Recently, it was demonstrated that ARCA-mRNA capping remarkably increased the expression level of reporter protein in mouse dendritic cells (Mockey et al. 2006).

In addition, the analogs we have developed that are resistant to enzymatic cleavage should be useful for biophysical, biochemical, and structural studies (e.g., NMR or X-ray crystallography) to provide new insights into the catalytic mechanism(s) of decapping enzymes and their regulation. The ability to obtain each S analog as diastereomerically pure populations offers the possibility of exploiting them as P-chiral probes useful for the elucidation of the stereochemical course of enzymatic processes involving cap as well as mapping of the active sites of cap binding proteins.

**FIGURE 10.** Interatomic contacts in the m^7GpppG -mouse eIF4E (28–217) complex, determined on the basis of the crystallographic structure (PDB entry 1L8B). The possible interactions are shown by dotted lines, and their lengths are expressed in Å. (Green) Carbon atoms, (blue) nitrogen, (yellow) phosphorus, (red) oxygen.

MATERIALS AND METHODS

Intermediate nucleotides were separated by ion-exchange chromatography on a DEAE-Sephadex A-25 (HCO_3^- form) column using a linear gradient of triethylammonium bicarbonate (TEAB) in deionized water and, after evaporation under reduced pressure with addition of ethanol, were isolated as triethylammonium salts. Final products (cap analogs) were separated by either analytical or semipreparative RP HPLC and, after repeated freeze-drying, were isolated as ammonium salts. Analytical HPLC was performed on a Spectra-Physics SP8800 apparatus, using a Supelcosil LC-18-T RP column (4.6×250 mm, flow rate 1.3 mL/min) with a linear gradient of 0%–25% methanol in 0.05 M ammonium acetate buffer (pH 5.9) and UV detection at 260 nm. Semipreparative HPLC was performed on a Waters 600E Multisolvant Delivery System apparatus, using a Waters HR-C-18 reverse phase column ($19 \text{ mm} \times 300 \text{ mm}$, flow rate 5.0 mL/min) with a linear gradient of methanol in 0.05 M ammonium acetate buffer (pH 5.9) and UV detection at 260 nm.

GMP and GDP were purchased from Sigma-Aldrich and converted into triethylammonium salts using Dowex 50 WX8 ion-exchange resin. Other nucleotides, i.e., m^7GMP , $\text{m}_2^{7,2'-\text{O}}\text{GMP}$, m^7GDP , and $\text{m}_2^{7,2'-\text{O}}\text{GDP}$, were prepared as previously reported (Darzynkiewicz et al. 1985; Jemielity et al. 2003). Thiophosphate triethylammonium salt was prepared from Na_3PSO_3 (Sigma) by conversion on Dowex 50 WX8 and, after evaporation to dryness

and re-evaporation with 99.8% ethanol, was stored at -20°C (Kowalska et al. 2007). 7-methylguanosine was prepared from guanosine by methylation with CH_3I similarly to previously reported procedure (Jones and Robins 1963). 7,2'-O-dimethylguanosine was obtained similarly from 2'-O-methylguanosine (Kusmierek and Shugar 1978).

The structure and homogeneity of each final product was confirmed by re-chromatography on RP HPLC, mass spectrometry using negative electrospray ionization (MS ESI⁻), and ^1H NMR and ^{31}P NMR spectroscopy (Table 4). ^1H NMR and ^{31}P NMR spectra were recorded at 25°C on a Varian UNITY-plus spectrometer at 399.94 MHz and 161.90 MHz, respectively. ^1H NMR chemical shifts were reported to sodium 3-trimethylsilyl-[2,2,3,3-D₄]-propionate (TSP) in D_2O as an internal standard. ^{31}P NMR chemical shifts were reported to 20% phosphorus acid in D_2O as an external standard. Mass spectra were recorded on a Micromass QToF 1 MS spectrometer.

General procedure for nucleotide imidazolidine derivatives (7, 8, and 12–15)

An appropriate nucleotide (1 eq., TEA salt), imidazole (8 eq.), and 2,2'-dithiodipyridine (3 eq.) were mixed in DMF (~ 2.5 mL/100 mg of nucleotide). Triethylamine (2 eq.) and triphenylphosphine (3 eq.) were added, and the mixture was stirred for 6–8 h. The product was precipitated from reaction mixture with a solution of

TABLE 4. ^1H NMR chemical shifts in parts per million (± 0.01) versus internal sodium 3-trimethylsilyl-[2,2,3,3- $^2\text{H}_4$]-propionate and ^{31}P NMR chemical shifts in parts per million (± 0.01) versus external H_3PO_4

	1a		1b		2a		2b		3a		3b	
	m^7G	G	m^7G	G	m^7G	G	m^7G	G	m^7G	G	m^7G	G
H8	— ^a	8.22	— ^a	8.14	9.00 ^b	8.04	9.01 ^b	7.94	9.11 ^b	8.01	9.08	8.01
H1'	5.92	5.85	5.91	5.84	5.83	5.74	5.84	5.74	5.92	5.79	5.90	5.79
H2'	4.58	4.62	4.58	4.62	4.58	4.71	4.45	4.60	4.58	4.69	4.54	4.67
H3'	4.46	4.47	4.46	4.47	4.49	4.54	4.42 ^c	4.42 ^c	4.50	4.49	4.49	4.42
H4'	4.35 ^c	4.35 ^c	4.35 ^c	4.35 ^c	4.27 ^c	4.36 ^c	4.36 ^c	4.39 ^c	4.34 ^c	4.39 ^c	4.36 ^c	4.42 ^c
H5'	4.38 ^c	4.31 ^c	4.38 ^c	4.31 ^c	4.42	4.27 ^c	4.39 ^c	4.22 ^c	4.38 ^c	4.27	4.37 ^c	4.29 ^c
H5''	4.26 ^c	4.31 ^c	4.26 ^c	4.31 ^c	4.36 ^c	4.27 ^c	4.36 ^c	4.20 ^c	4.33 ^c	4.26	4.35 ^c	4.29
CH_3 (N7)	4.07	—	4.05	—	4.06	—	4.03	—	4.07	—	4.07	—
$\text{P}\alpha$	44.17	—	44.17	—	—12.37	—	—12.37	—	—11.26	—	—11.26	—
$\text{P}\beta$	—23.86	—	—23.86	—	30.27	—	30.18	—	—23.79	—	—23.79	—
$\text{P}\gamma$	—11.29	—	—11.29	—	—12.37	—	—12.37	—	43.66	—	43.26	—
	4a		4b		5a		5b		6a		6b	
	$\text{m}_2^{7,2'-\text{O}}\text{G}$	G	$\text{m}_2^{7,2'-\text{O}}\text{G}$	G	$\text{m}_2^{7,2'-\text{O}}\text{G}$	G	$\text{m}_2^{7,2'-\text{O}}\text{G}$	G	$\text{m}_2^{7,2'-\text{O}}\text{G}$	G	$\text{m}_2^{7,2'-\text{O}}\text{G}$	G
H8	— ^a	8.10	— ^a	8.07	9.01 ^b	8.03	— ^a	8.10	— ^a	8.07	9.01 ^b	8.03
H1'	5.94	5.81	5.93	5.80	5.97	5.80	5.94	5.81	5.93	5.80	5.97	5.80
H2'	4.26	4.68	4.21	4.66	4.24	4.68	4.26	4.68	4.21	4.66	4.24	4.68
H3'	4.56	4.50	4.52	4.48 ^c	4.54	4.49	4.56	4.50	4.52	4.48 ^c	4.54	4.49
H4'	4.30 ^c	4.37 ^c	4.33 ^c	4.35 ^c	4.33 ^c	4.27 ^c	4.30 ^c	4.37 ^c	4.33 ^c	4.35 ^c	4.33 ^c	4.27 ^c
H5'	4.39 ^c	4.30 ^c	4.46 ^c	4.28 ^c	4.41	4.30 ^c	4.39 ^c	4.30 ^c	4.46 ^c	4.28 ^c	4.41	4.30 ^c
H5''	4.30 ^c	4.30 ^c	4.34 ^c	4.26 ^c	4.32 ^c	4.27 ^c	4.30 ^c	4.30 ^c	4.34 ^c	4.26 ^c	4.32 ^c	4.27 ^c
CH_3 (N7)	4.08	—	4.07	—	4.06	—	4.08	—	4.07	—	4.06	—
CH_3 (2'-O)	3.59	—	3.59	—	3.60	—	3.59	—	3.59	—	3.60	—
$\text{P}\alpha$	43.61	—	43.70	—	—12.10	—	—12.10	—	—11.25	—	—11.32	—
$\text{P}\beta$	—23.86	—	—23.80	—	30.33	—	30.23	—	—23.85	—	—23.72	—
$\text{P}\gamma$	—11.33	—	—11.34	—	—12.10	—	—12.10	—	43.63	—	43.13	—

^aExchangeable protons.

^bExchangeable but visible protons.

^cApproximate value because of signal overlapping.

anhydrous NaClO_4 (1 eq. per one negative charge) in dry acetone (~ 8 mL/1 mL of DMF). After cooling at 4°C , the precipitate was filtered, washed repeatedly with cold, dry acetone, and dried in vacuum over P_4O_{10} . Yields were 80%–100%. In the case of $m^7\text{GMP}$, due to its poorer solubility in DMF, a twofold larger excess of reagents was used, and reaction time was extended to 24 h.

General procedure for nucleoside 5'-O-phosphorothioates (9–11)

A suspension of an appropriate nucleoside (1 eq., dried overnight in vacuum over P_4O_{10}) in trimethyl phosphate (1.5 mL/100 mg of nucleoside) was cooled to 0°C in ice/water bath. 2,6-dimethylpyridine (3 eq.) and PSCl_3 (1.5 eq.) were added. The reaction was maintained overnight at 0°C , then quenched with 10 volumes of water and stirred for 1 h at room temperature. The product was separated by DEAE Sephadex chromatography using a linear gradient of 0–0.7 M TEAB. Yields: 380 mg of **9** (0.67 mmol) starting from 257 mg (0.91 mmol) of guanosine (74%); 57 mg of **10** (0.10 mmol) starting from 120 mg (0.42 mmol) of 7-methylguanosine (24%); 75 mg of **11** (0.13 mmol) starting from 70 mg (0.23 mmol) of 7,2'-O-dimethylguanosine (53%).

Synthesis of nucleoside 5'-O-(2-thiodiphosphates)

7,2'-O-dimethylguanosine 5'-O-(2-thiodiphosphate) (**17**)

To a suspension of **14** (100 mg, 0.21 mmol) and thiophosphate triethylammonium salt (220 mg) in 5 mL of DMF, anhydrous ZnCl_2 (190 mg, 1.40 mmol) was added. The resulting solution was stirred for 20 min at room temperature. The reaction was quenched by addition of EDTA (520 mg, 1.40 mmol) in 50 mL of water and quickly neutralized with solid NaHCO_3 . The product was isolated on DEAE Sephadex using a 0–1.0 M gradient of TEAB. Yield: 106 mg (0.15 mmol) of **17** as TEA salt (71%).

7-methylguanosine 5'-O-(2-thiodiphosphate) (**16**)

This compound was synthesized as described for **17** starting from **13** (40 mg, 0.089 mmol) and thiophosphate triethylammonium salt (100 mg). Yield: 31 mg (0.046 mmol) of **16** as TEA salt (52%).

Synthesis of cap 5 analogs

$m^7\text{Gppp}_s\text{G}$ D1 and D2 (**1a** and **1b**)

To a suspension of **9** (10 mg, 0.018 mmol) and **7** (15 mg, 0.027 mmol) in DMF (0.8 mL), anhydrous ZnCl_2 (30 mg, 0.22 mmol) was added. The reaction was maintained for 2 d at room temperature and then quenched by addition of 90 mg of EDTA in 10 mL of water and neutralized with solid NaHCO_3 . The diastereomers **1a** and **1b** were separated by analytical RP HPLC. Yield: 0.8 mg of **1a** and 1.0 mg of **1b** as NH_4^+ salts.

$m^7\text{Gppp}_s\text{G}$ D1 and D2 (**2a** and **2b**)

Synthesized and separated as described for **1a** and **1b** starting from **16** (20 mg, 0.030 mmol), **15** (23 mg, 0.053 mmol), and ZnCl_2 (60 mg, 0.44 mmol) in 2 mL of DMF. Yield: 2.2 mg of **2a** and 1.8 mg of **2b** as NH_4^+ salts.

$m^7\text{Gp}_s\text{ppG}$ D1 and D2 (**3a** and **3b**)

Synthesized as described for **1a** and **1b** starting from **10** (58 mg, 0.090 mmol), **12** (120 mg, 0.22 mmol), and ZnCl_2 (249 mg, 1.8 mmol) in 3.5 mL of DMF. Yield: 14.7 mg of **3a** and 10.1 mg of **3b** as NH_4^+ salts.

$m_2^{7,2'}\text{-O-Gppp}_s\text{G}$ D1 and D2 (**4a** and **4b**)

Compounds **9** (48 mg, 0.084 mmol) and **8** (57 mg, 0.10 mmol) were suspended in 2 mL of DMF. Subsequently, anhydrous ZnCl_2 (115 mg, 0.84 mmol) was added. The resulting solution was maintained for 2 d at room temperature. The reaction was quenched by addition of 350 mg of EDTA in 30 mL of water and neutralized with solid sodium bicarbonate. Products were separated by semipreparative RP HPLC using a linear gradient of methanol in 0.05 M ammonium acetate, pH 5.9, from 0%–25% within 45 min. Yield: 5.2 mg of **4a** and 7.4 mg of **4b** as NH_4^+ salts.

$m_2^{7,2'}\text{-O-Gpp}_s\text{pG}$ D1 and D2 (**5a** and **5b**)

Synthesized as described for **4a** and **4b** starting from **17** (106 mg, 0.16 mmol), **15** (103 mg, 0.24 mmol), and ZnCl_2 (260 mg, 1.9 mmol) in 5 mL of DMF. The reaction was quenched with 800 mg of EDTA in 100 mL of water and neutralized with solid sodium bicarbonate. Products were separated by semipreparative RP HPLC using isocratic 0.05M ammonium acetate, pH 5.9. Yield: 10.0 mg of **5a** and 12.1 mg of **5b** as NH_4^+ salts.

$m_2^{7,2'}\text{-O-Gp}_s\text{ppG}$ D1 and D2 (**6a** and **6b**)

This compound was synthesized as described for **4** starting from **11** (70 mg, 0.15 mmol), **12** (107 mg, 0.20 mmol), and anhydrous ZnCl_2 (220 mg, 1.6 mmol) in 3 mL of DMF. The reaction was quenched with 650 mg of EDTA in 70 mL of water. Products were separated by semipreparative RP HPLC using a linear gradient of methanol in 0.05 M ammonium acetate, pH 5.9, from 0%–50% within 45 min. Yield: 15 mg of **6a** and 20 mg of **6b** as NH_4^+ salts.

eIF4E fluorescence binding titration

Mouse eukaryotic initiation factor eIF4E (residues 28–217) was expressed in BL21(DE3) *Escherichia coli* strain. The protein was purified from inclusion bodies, and guanidinium-solubilized protein was refolded by one-step dialysis and purified by ion-exchange chromatography on a HiTrapSP column without contact with cap analogs. The concentration of eIF4E was determined spectrophotometrically ($\epsilon_{280} = 53\,400\text{ cm}^{-1}\text{M}^{-1}$).

Fluorescence titration measurements were carried out on an LS-50B spectrofluorometer (Perkin Elmer Co.) in 50 mM HEPES/KOH (pH 7.2), 100 mM KCl, 0.5 mM EDTA, 1 mM DTT at $20.0 \pm 0.2^\circ\text{C}$. Aliquots of 1 μL increasing concentration of cap analog solutions were added to 1.4 mL of 0.1 μM protein solutions. Fluorescence intensities (excitation at 280 nm with 2.5-nm bandwidth and detection at 337 nm with 4-nm bandwidth and 290 nm cutoff filter) were corrected taking into account sample dilution and the inner filter effect. Equilibrium association constants (K_{AS}) were determined by fitting the theoretical dependence of fluorescence intensity on the total concentration of cap

analog to the experimental data points, according to equation described previously (Niedzwiecka et al. 2002). The concentration of protein was fitted as a free parameter of equilibrium equation showing amount of “active” protein. The final K_{AS} was calculated as a weighted average of 3–10 independent titrations, with the weights taken as the reciprocals of the numerical standard deviations squared. Numerical nonlinear least-squares regression analysis was performed using ORIGIN 6.0 (Microcal Software Inc.).

The Gibbs free energy of binding was calculated from the K_{AS} value according to the standard equation $\Delta G^\circ = -RT \ln K_{AS}$.

Enzymatic hydrolysis of S analogs with human and *C. elegans* DcpS

Human and nematode DcpS were expressed in *E. coli* according to the procedures described previously (Cohen et al. 2004). Both proteins were stored at -80°C in 20 mM Tris buffer, pH 7.5, containing 50 mM KCl, 0.2 mM EDTA, 1 mM DTT, 0.5 mM PMSF, and 20% glycerol. An appropriate cap analog at 4 μM concentration was treated with 5.0 or 7.0 μL of DcpS (from human or *C. elegans*, respectively) in 500 μL of 50 mM Tris buffer, pH 7.9, containing 20 mM of MgCl_2 and 60 mM of $(\text{NH}_4)_2\text{SO}_4$ at 37°C for 60–90 min. Every 15–20 min, a 100- μL sample was collected from the reaction mixture and deactivated by incubation for 3 min at 90°C . Collected samples were analyzed without further treatment by analytical RP HPLC using a linear gradient of methanol in 0.1 M KH_2PO_4 , pH 6.0, from 0%–50% within 15 min and UV detection at 260 nm.

Inhibition of cap-dependent translation

A micrococcal nuclease-treated rabbit reticulocyte lysate was used for in vitro translation (Cai et al. 1999). Optimal cap-dependent translation was achieved at 100 mM potassium acetate and 1.4 mM magnesium chloride. To determine inhibition of translation by various cap analogs, natural rabbit globin mRNA was added to the lysate at 5 $\mu\text{g}/\text{mL}$, and protein synthesis was measured by incorporation of $[^3\text{H}]\text{Leu}$. Normalization of K_I data was performed as described previously (Cai et al. 1999). The concentrations of dinucleotide cap analog solutions were measured by UV absorbance at pH 7.0 using $\lambda = 255 \text{ nm}$ and $\epsilon_M = 22.6 \times 10^{-3} \text{ M}$.

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