Mathematical Model of Nitrogen Metabolism in Yeast

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Modeling with Yeast

- Model
 - Define all state variables, terms, and parameters involved in the system of differential equations
 - Analysis of steady-state
 - Simulation of dynamics
- Model in relation with articles

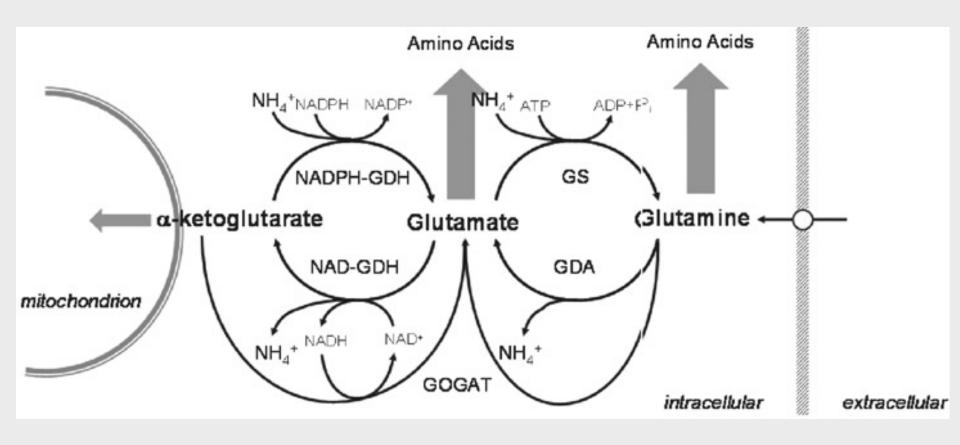
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Background

- The article "The Concentration of Ammonia Regulates Nitrogen Metabolism in Saccharomyces cerevisiae" provides numerous models outlining the effects that changes in ammonia concentration have on the nitrogen metabolism of yeast.
- To build a better understanding of these processes, a model was created off of a list of equations gathered from existing data. The models were then compared to those featured in the paper described above.
- The article "Nitrogen regulated transcription and enzyme activities in continuous cultures of Saccaromyces cerevisiae" provides additional information on the activities of yeast in continuous cultures of difference dilution rates under nitrogen limitation.
- This article studied the changes in carbon and nitrogen fluxes under such environments and the effect different dilution rates had on transcription and nitrogen regulating genes.

Central Nitrogen Metabolism Pathway



van Riel & Sontag (2006) IEEE Proc. -Syst. Biol. 153: 263-274

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Differential Equations and Variables

$$\mathbf{A} \xrightarrow{r_1} \mathbf{B}$$

$$\mathbf{B} \overset{r_2}{\longleftrightarrow} \mathbf{C}$$

$$A \stackrel{r_1}{\longleftrightarrow} B \qquad B \stackrel{r_2}{\longleftrightarrow} C \qquad A + C \stackrel{r_3}{\longrightarrow} 2B$$

State Variables

 $a = \alpha$ -ketogluterate

b = glutamate

c = glutamine

System of differential equations

$$\frac{da}{dt} = -r_1 a + r_{-1} b - r_3 ac$$

$$\frac{db}{dt} = r_1 a - r_{-1} b + r_{-2} c - r_2 b + 2r_3 ac$$

$$\frac{dc}{dt} = -r_{-2} c + r_2 b - r_3 ac$$

Steady-state Analysis

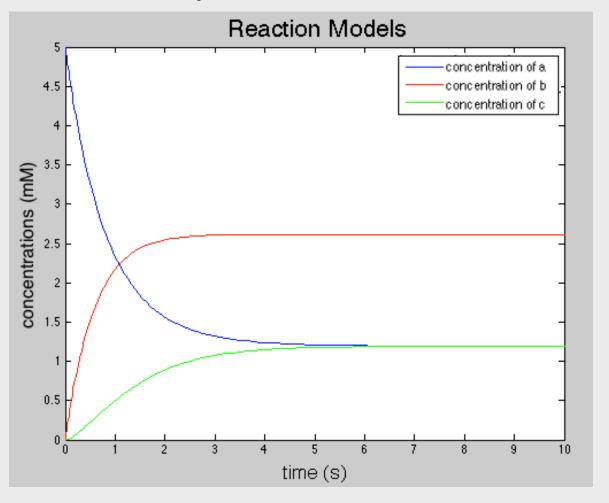
At
$$da/dt = 0$$
, if $b = 0$
 $c = (r_{-1}b / r_3a) - r_1/r_3$
At $db/dt = 0$ and $dc/dt = 0$,
 $c = r_2b / (r_{-2} + r_3a)$

- r_1/r_3

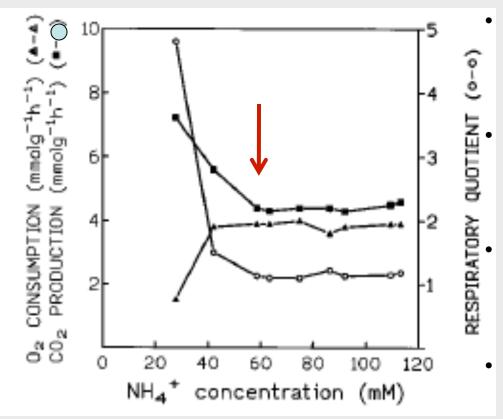
b/a

Graphical Output of the Dynamic's Simulation

All parameters equal to 1



When the Culture Was Switched from Ammonium Limitation to Ammonium Excess, No Further Changes in Respiratory Quotient Were Observed



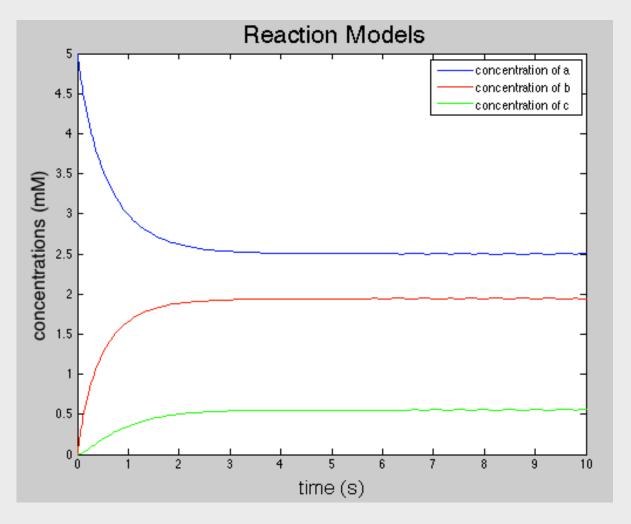
The "respiratory quotient" shows whether the culture is performing fermentation or respiration.

- At 29 mM NH₄⁺ in the feed, the culture is performing fermentation.
- Beginning at 44 mM NH₄⁺, the culture switches over to respiration.
- After 61 mM NH₄⁺ (excess), no further changes in carbon metabolism are observed.

respiratory quotient = CO₂ produced/O₂ consumed

Modifying r₋₁ parameter

Set $r_{-1} = 2$



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Discussion

"If the ammonia concentration is the regulator, this may imply that S. cerevisiae has an ammonia sensor which could be a two-component sensing system for nitrogen..."

- The two component system could be referring to the two step reaction that occurs in the yeast. The first component is the reaction transforming α-ketoglutarate into glutamae and then the second component is the reaction transforming glutamate into glutamine.
- Before each component, the nitrogen concentration must be evaluated before proceeding with the reaction. Therefore, an ammonia sensor would have to remain sensitive to both instances.
- The concept of two components could also be referring to the extracellular and intracellular concentrations of nitrogen, which both provide a unique measure. For the hypothetical ammonia sensor to run, both concentrations must be considered.

Summary

- Modeling with Yeast
- Model
 - State variables
 - System of Differential Equations
 - Terms
 - Parameters
 - Analysis of steady-state
 - Simulation of dynamics with graphical output
- Model in relation with articles

Work Cited

- ter Schure, E.G., Sillje, H.H.W., Verkleij, A.J., Boonstra, J., and Verrips, C.T. (1995) Journal of Bacteriology 177: 6672-6675.
- Schure, E. G. Ter, H. H. W. Sillje, L. J. R. M. Raeven, J. Boonstra, A. J. Verkleij, and C. T. Verrips. (1995) *Microbiology* 141.5: 1101-108.